

DATA AND OBSERVATIONS

1. Suggestion for your data table:

Trial	Equivalence point (mL)
1	17 mL
2	16.25 mL

2. Make sure to record all observations.

RESULTS AND CALCULATIONS

1. Calculate the $[\text{OH}^-]$ from the results of your titrations. Explain your calculations.
2. Calculate the $[\text{Ca}^{2+}]$. Explain your calculations.
3. Calculate the K_{sp} for calcium hydroxide. Explain your calculations.
4. Find the accepted value of the K_{sp} for calcium hydroxide and compare it with your value. Discuss the discrepancy and suggest possible sources of experimental error. Comment on the error and results.
5. Indicate any changes that you would make to the procedure.

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